

6-endo-dig Free-Radical Carbocyclizations: A New Strategy for the Synthesis of Cyclitols

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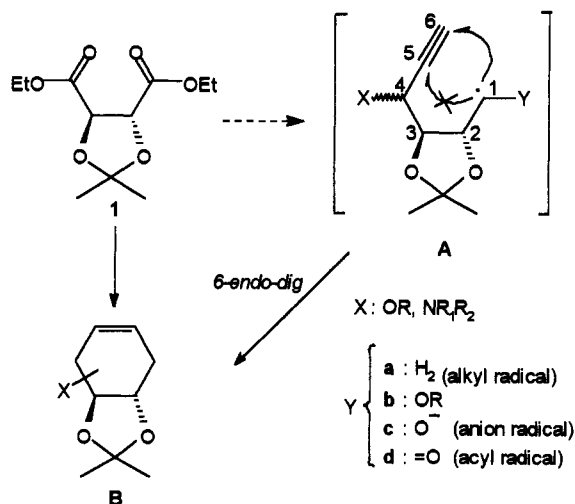
Received November 12, 1993*

Summary: The 6-endo-dig free radical carbocyclization of the chiral 1-iodo-5-alkynes **5** and **6** has been used to prepare the cyclitols **7-9**.

Cyclitols are polyhydroxycyclohexane derivatives with important biological activities.¹ There is currently considerable interest in the synthesis of these types of compounds.² We have recently described a new approach for the preparation of aminodeoxyinositols³ and branched-chain cyclitols⁴ by 6-exo-trig radical cyclization strategies.

We now report a novel strategy for the asymmetric synthesis of inositols. This method is based in the 6-endo-dig⁵ free radical carbocyclization⁶ of chiral, conveniently functionalized 1-iodo-5-alkynes.⁷ We hypothesized that, as the 5-exo-dig ring closure of intermediate **A** would be disfavored because two cyclopentanes cannot be *trans* fused, the cyclization of radical species **A** could be exclusively directed, by the 6-endo-dig path, to the functionalized cyclohexene **B** (Scheme 1). Although kinetically controlled radical cyclizations follow preferentially the *exo* mode,^{5a} major *endo* products have been observed in the ring closure with sterically hindered *exo* atoms,^{5b} in the annelation of amidyl radicals with terminal bonds,⁸ and when a silicon atom is involved in the radical.⁹

Scheme 1



It has been also claimed that the “*endo* ring closure is excluded for simple 5-hexynyl radicals...and when closure involves a terminal C=X bond.”¹⁰

Contrary to these expectations we now report for the first time the successful 6-endo-dig free-radical-promoted carbocyclization of simple 5-hexynyl radicals. We have shown that the subtle *trans* located 1,3-dioxolane moiety in a radical precursor can influence the regioselectivity in the carbon-centered cyclization. We have used this concept in the transformation of (2*R*,3*R*)-diethyl 2,3-*O*-isopropylidene tartrate **1**,¹¹ via species **Aa**¹² (Scheme 1), into cyclitols **7-9** (Scheme 2).

These ideas have been put into practice with the synthesis and cyclization of compounds **5** and **6**. The simple route **1**¹¹ → **2**¹³ → **3**,¹⁴ followed by Swern oxidation¹⁵ and ethynylmagnesium bromide addition to the aldehyde **4** (Scheme 2), allowed us to obtain the radical precursor **5** in four steps and good overall yield. Compound **5** was isolated as a mixture of *anti/syn* isomers^{16,17} at C4 in a

* Abstract published in *Advance ACS Abstracts*, March 1, 1994.

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(12) The present approach allows the preparation of epimers at C2/C3 in compounds of type **B** (Scheme 1) by using the readily available (*D*)-**1**. The different and selective manipulation of these starting materials should also provide an easy entry to other interesting radical species (**Bb-d**, Scheme 1).

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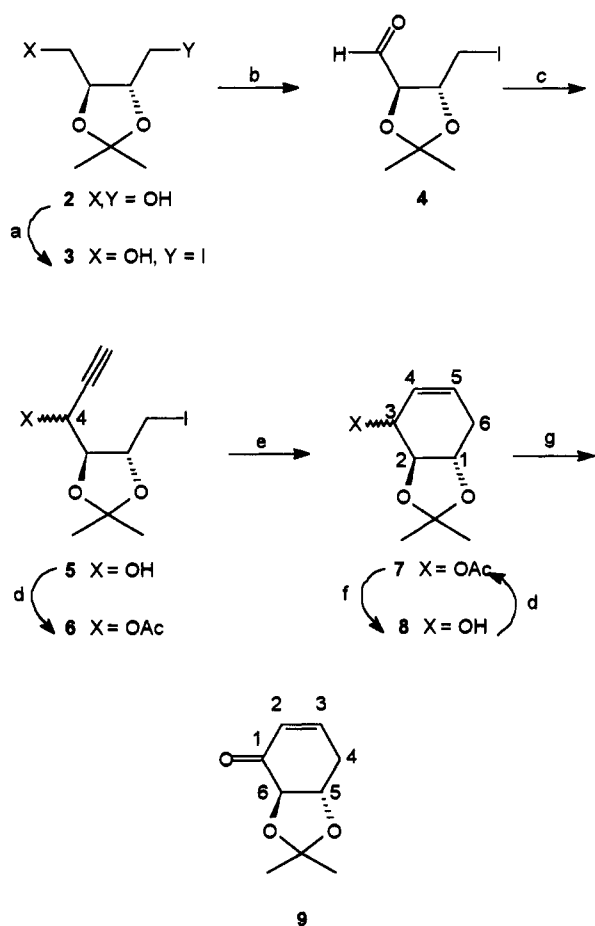
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(16) The presence of boron-trifluoride etherate during the addition of the Grignard reagent to compound **4** did not improve the *anti/syn* ratio.

(17) The assignment of relative configurations (*anti/syn*) in compounds **5** and **6** has been made by comparison of the chemical shifts (δ), in the ¹H NMR spectra, with those of analogous products obtained in the additions of acetylenic reagents to similar tartraldehide derivatives (Mukaiyama, T.; Suzuki, K.; Yamada, T.; Tabusa, F. *Tetrahedron* 1990, 46, 265) and have been confirmed by the analysis of spectral data of the cyclized products. Ratios of the (*anti/syn*) isomers were determined by ¹H NMR integration of the crude mixtures.

Scheme 2



Reagents: (a) Ph_3P , I_2 , imidazole, toluene (59%); (b) $(\text{CO}_2)_2\text{Cl}_2$, DMSO, Et_3N , CH_2Cl_2 (79%); (c) $\text{BrMgC}\equiv\text{CH}$, THF (84%); (d) Ac_2O , pyridine (90%); (e) AIBN, Bu_3SnH , toluene; (f) MeOH, Et_3N , H_2O (65%); (g) PDC, CH_2Cl_2 (70%).

63:37 ratio. This unseparable mixture was submitted to standard acetylation conditions giving a mixture of acetates 6 (*anti/syn* = 63/37) that we were also unable to separate.

With the radical precursors in hand, the cyclization was attempted.¹⁸ The tributyltin hydride + AIBN-mediated free radical cyclization of acetate 6 (*anti* + *syn*) gave the 6-*endo-dig* product 7; after careful flash chromatography,¹⁹ we could isolate 7, as a mixture of isomers at C3

(18) Cyclization of 5 or 6: In a typical experiment, to a solution of the radical precursor in toluene (0.02 M), under argon and at reflux, was slowly added a solution of tributyltin hydride + AIBN (cat.) in toluene dropwise *via* syringe pump over 7 h. The mixture was cooled and the solvent evaporated. The residue was dissolved in ether and treated with an aqueous solution (20%) of potassium fluoride. After the mixture was stirred overnight, the organic phase was separated, dried, evaporated, and submitted to chromatography, eluting with hexane/ethyl acetate mixtures. Ratios of the purified cyclization compounds determined by ^1H NMR integration were similar to the values determined in crude mixtures.

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(*S/R* \approx 1/1), and traces of the more polar isomer, pure 7 (C3 *S*), in 66% total yield. Basic hydrolysis (MeOH, Et_3N , H_2O) of compound 7 (C3 *S*) gave 8 (C3 *S*) in 88% yield. The ^1H NMR analysis of 7 (C3 *S*) and 8 (C3 *S*) unequivocally established the relative stereochemistry at C2/C3. Thus, since the molecule is in a rigid conformation, the value of $J_{2,3} = 3.8$ Hz for $^1\text{H}_2$ in 7 (C3 *S*) suggests an axial-equatorial arrangement of protons H2 and H3. In addition, significant cross-peaks (H2/H3, H3/H6ax, H2/H6ax, and H1/H6eq) in the NOESY spectrum of 7 (C3 *S*) indicate a quasiaxial orientation of the OH group.

The cyclization of compound 5 (*anti* + *syn*), under similar conditions, and after flash chromatography,¹⁹ afforded a mixture of isomers 8 at C3 (*S/R*:45/55) in 50% yield. We were unable to separate them by chromatography. As shown in the ^1H NMR spectrum and by comparison with the spectral data of 8 (C3 *S*), this compound was the major isomer in the mixture of epimers 8 and, as expected, corresponded to the ring closure of the major 5 (*anti*) precursor.¹⁷

Finally, simple acetylation or PDC oxidation²⁰ of the allylic alcohols 8 gave the acetates 7 (C3 *S/R* = 55/45; 90% yield) and the cyclohexenone 9 (70% yield), respectively. Thus, in spite of the low diastereomeric excess obtained in the formation of alcohol 5, we could transform the corresponding cyclized isomers 8 into the homogeneous ketone 9, a highly functionalized, chiral intermediate for the synthesis of different cyclitols and natural products.²¹

In summary, we have shown that the 6-*endo-dig* cyclization of simple 5-hexynyl radicals derived from precursors 5 and 6 is feasible and the corresponding carbocycles can be obtained in good yields. This results in a new strategy for the synthesis of inositols. Work is now in progress in order to apply this methodology to other precursors and will be reported in due course.

Acknowledgment. This work was supported by DGI-CYT (PB 90-0078) and Comunidad de Madrid (C195/91A).

Supplementary Material Available: Experimental procedures for the preparation of compounds 3 and 5 and spectral data of compounds 3, 5, 6, 7 (C3 *S*), 8 (C3 *S*), and 9 (2 pages). This material is contained in libraries on microfiche, immediately follows this article in the microfilm version of the journal, and can be ordered from the ACS; see any current masthead page for ordering information.

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(21) (1*S*,2*R*,3*S*)-2,3-*O*-Isopropylidencyclohex-4-ene-1,2,3-triol, an isomer of compound 8 (C3 *S*), has been recently synthesized and transformed into pseudosugars (Dumortier, L.; Van der Eycken, J.; Vandewalle, M. *Synlett* 1992, 245. See also: Maycock, C. D.; Barros, M. T.; Santos, A. G.; Godinho, L. S. *Tetrahedron Lett.* 1992, 33, 1633). In addition, the C1/C3 epimer of compound 8 (C3 *S*) has also been prepared in enantiomerically pure form by enzymatic resolution and transformed into the C5 epimer of product 9; this ketone is the key intermediate in the total synthesis of (+)-palitantin (Deruythere, X.; Dumortier, L.; Van der Eycken, J.; Vandewalle, M. *Synlett* 1992, 51).